

## Even More Naming Ionic Compounds

Name the following ionic compounds:

- 1)  $\text{CuSO}_4$  \_\_\_\_\_
- 2)  $\text{Na}_2\text{O}$  \_\_\_\_\_
- 3)  $\text{Pb}(\text{OH})_2$  \_\_\_\_\_
- 4)  $\text{LiC}_2\text{H}_3\text{O}_2$  \_\_\_\_\_
- 5)  $\text{Cu}_3(\text{PO}_4)_2$  \_\_\_\_\_
- 6)  $\text{NH}_4\text{Br}$  \_\_\_\_\_
- 7)  $\text{CaCO}_3$  \_\_\_\_\_
- 8)  $\text{KCN}$  \_\_\_\_\_
- 9)  $\text{FeBr}_3$  \_\_\_\_\_
- 10)  $\text{Co}(\text{NO}_2)_2$  \_\_\_\_\_

Write the formulas of the following compounds:

- 11) manganese (IV) nitride \_\_\_\_\_
- 12) aluminum sulfide \_\_\_\_\_
- 13) magnesium selenide \_\_\_\_\_
- 14) chromium (III) sulfate \_\_\_\_\_
- 15) silver phosphate \_\_\_\_\_
- 16) palladium (IV) bromide \_\_\_\_\_
- 17) titanium (II) arsenide \_\_\_\_\_
- 18) vanadium (V) telluride \_\_\_\_\_
- 19) beryllium oxide \_\_\_\_\_
- 20) nickel (III) carbonate \_\_\_\_\_