## **Moles Homework**

Honors Chemistry

- 1) How many grams are there in 4.5 moles of copper (II) chloride?
- 2) How many moles are there in  $4.5 \times 10^{24}$  molecules of boron trichloride?
- 3) How many molecules are there in 45 moles of diphosphorus trioxide?
- 4) How many molecules are there in 23 grams of sulfur dioxide?
- 5) How many ions are present in 83 grams of ammonium nitrate?
- 6) How many molecules are in 450 grams of oxygen?
- 7) How many grams are in  $1.0 \times 10^{24}$  molecules of SiO<sub>2</sub>?
- 8) How many molecules are there in 2280 grams of methane?

9) Using your knowledge of mole calculations and unit conversions, determine how many atoms there are in 1 gallon of gasoline. Assume that the molecular formula of gasoline is  $C_6H_{14}$  and that the density of gasoline is approximately 0.85 grams/mL.