Product Formation Homework

For each of the problems below, indicate whether you believe a reaction will occur. For those that you believe will occur, write the expected products of the reactions. Remember to write symbols for the states of each compound.

1)
$$_{\text{____}} H_2SO_4 + _{\text{____}} MgCl_2 \rightarrow$$

3)
$$Mg(OH)_2 + MCI \rightarrow$$

4) Mg +
$$___$$
 S₈ \rightarrow

5)
$$___$$
 SrBr₂ + $___$ (NH₄)₃PO₄ \rightarrow

6)
$$H_2CO_3 \rightarrow$$

- 7) Your teacher has told you to produce magnesium hydroxide by combining lithium hydroxide with magnesium nitrate in a double replacement reaction.
 - a) Write the balanced equation for this reaction.
 - b) Write the complete ionic equation for this reaction.
 - c) Write the net ionic equation for this reaction.
- 8) Your teacher has told you to produce lithium nitrate using the reaction above. Explain how you can do this.