Properties of Metals

1) What are delocalized electrons?

2) Why are metals bendy?

3) Why do metals conduct electricity?

4) Very tall buildings are usually made of concrete (which is an ionic compound) that is reinforced/made stronger with steel bars. Based on what you know of both ionic compounds and metals, why is this construction method a good idea?

5) We have talked about why metals are malleable and ductile. Considering the structure of metals, why aren't metals even more bendy (like a cooked noodle, where it can bend a very great deal)? Explain your answer.